



Swansea University
Prifysgol Abertawe



TATA STEEL

Aesthetic and performance enhancements of ZMA coated steels

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EPSRC

Engineering and Physical Sciences
Research Council



Research Fund for Coal & Steel

MicroCorr



Llywodraeth Cymru
Welsh Government

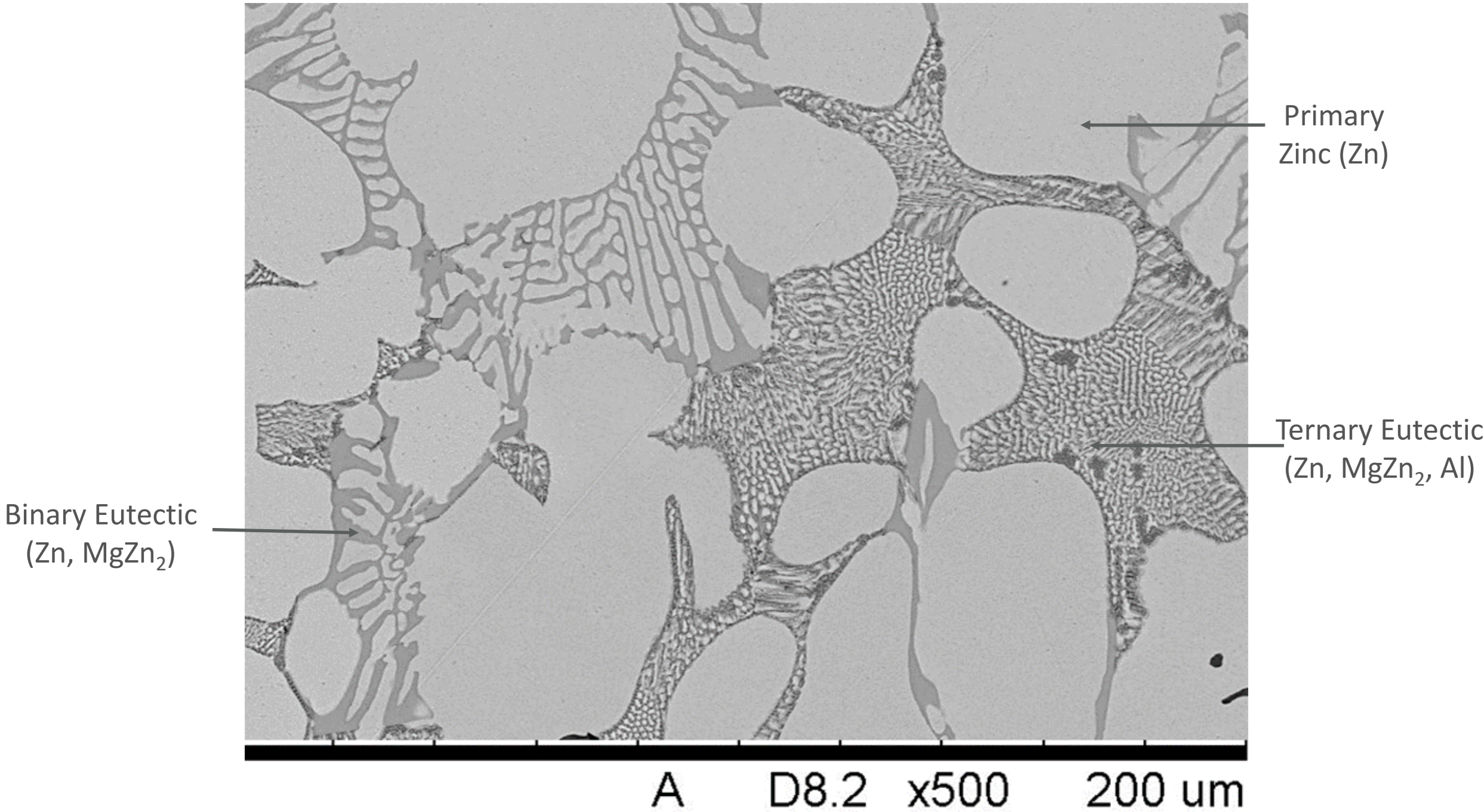
**Cronfa Gymdeithasol Ewrop
European Social Fund**

Zinc Magnesium Aluminium (ZMA):

- Provides cathodic protection
- Acts as barrier



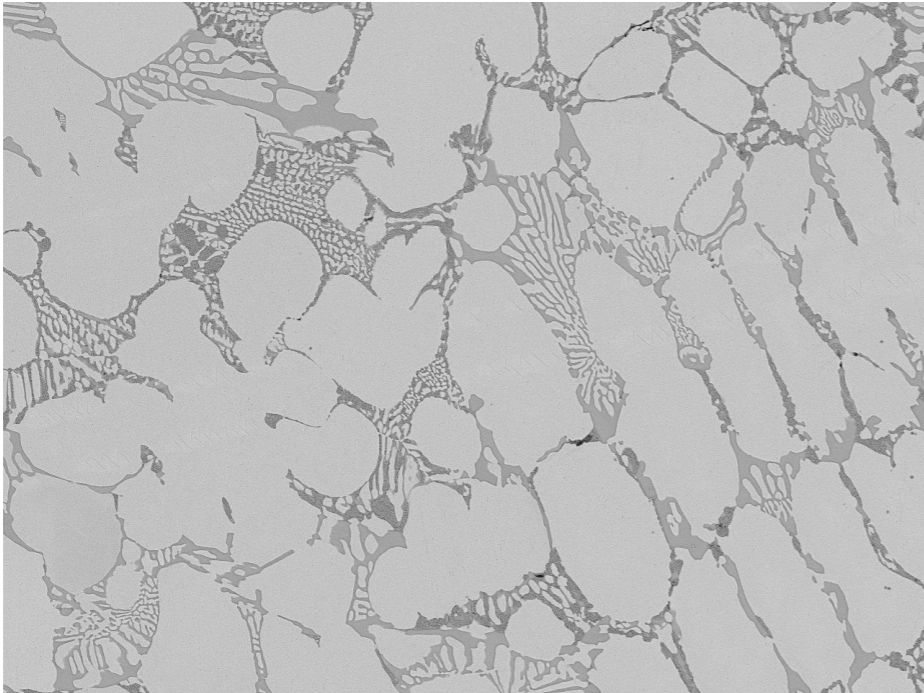
Microstructure of ZMA:



Zn - (1-2) wt.% Al - (1-2) wt.% Mg

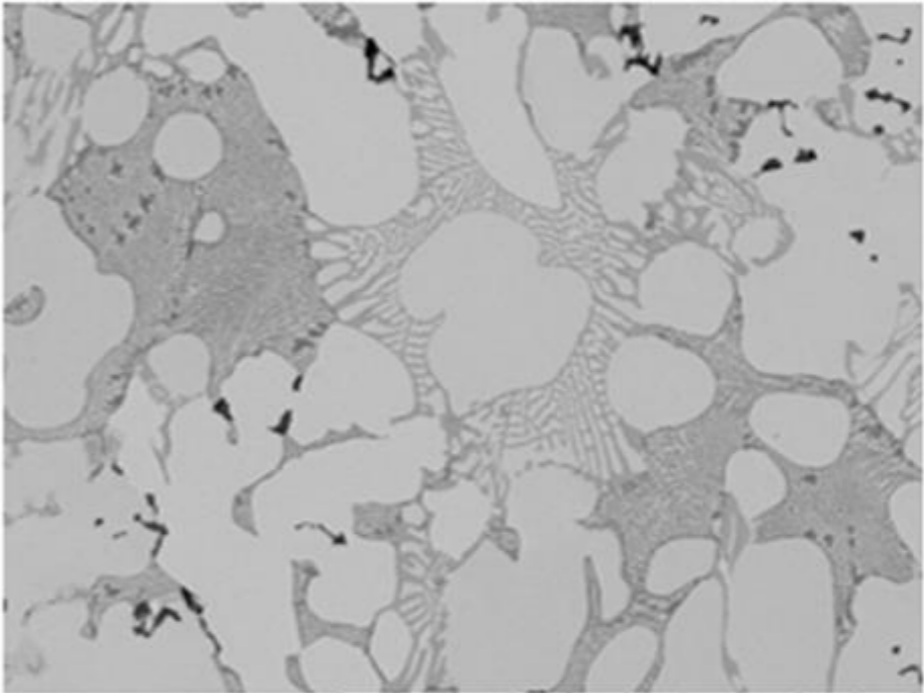
Experimental samples:

0 Ge

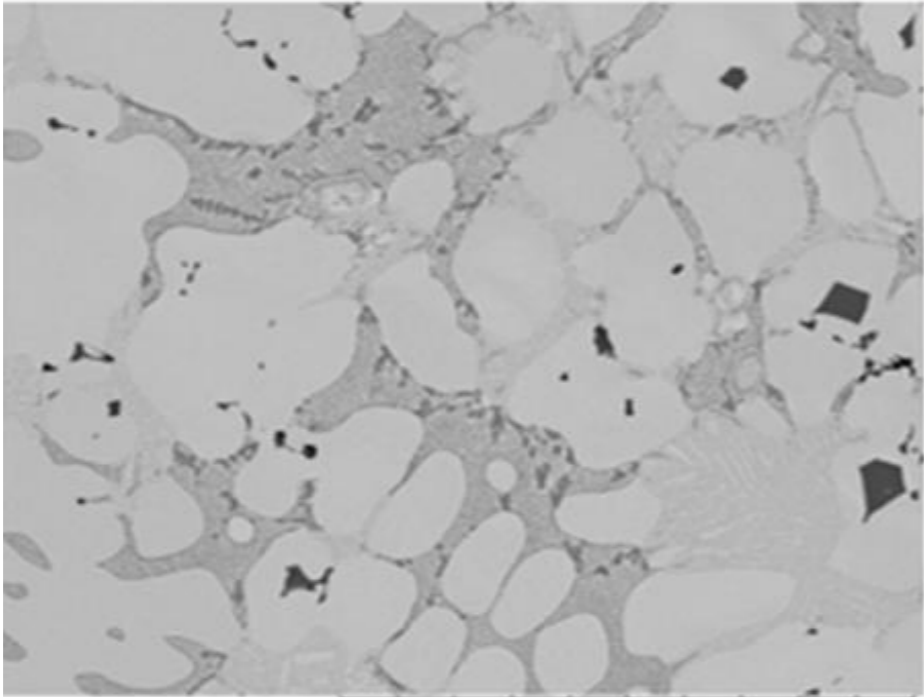


2018/08/09 12:19 A D9.2 x500 200 um

0.19 Ge

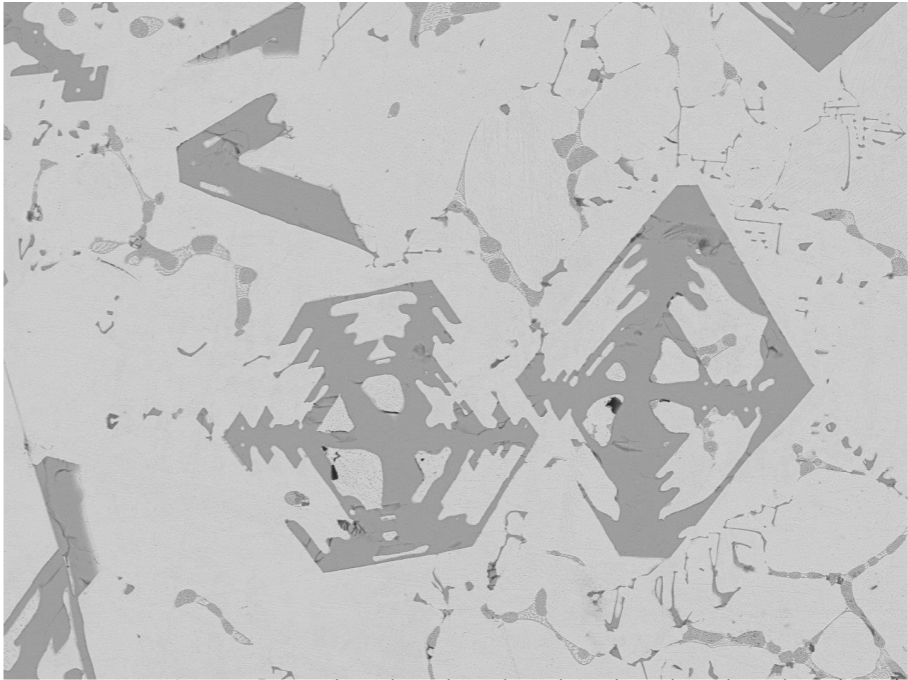


A D8.2 x500 200 um



A D8.2 x500 200 um

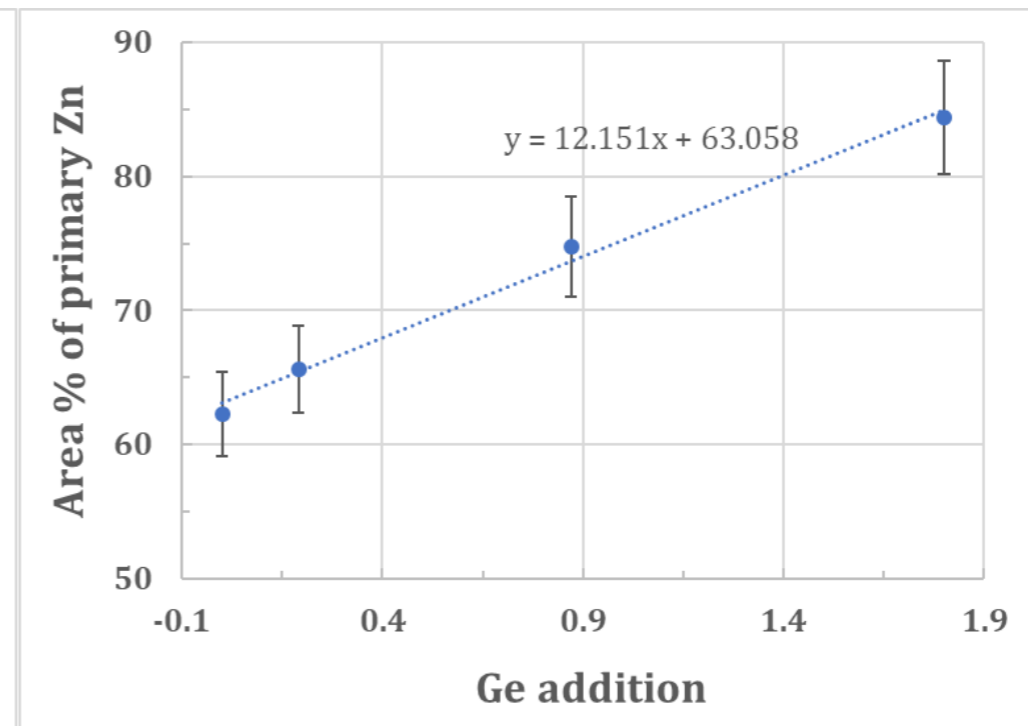
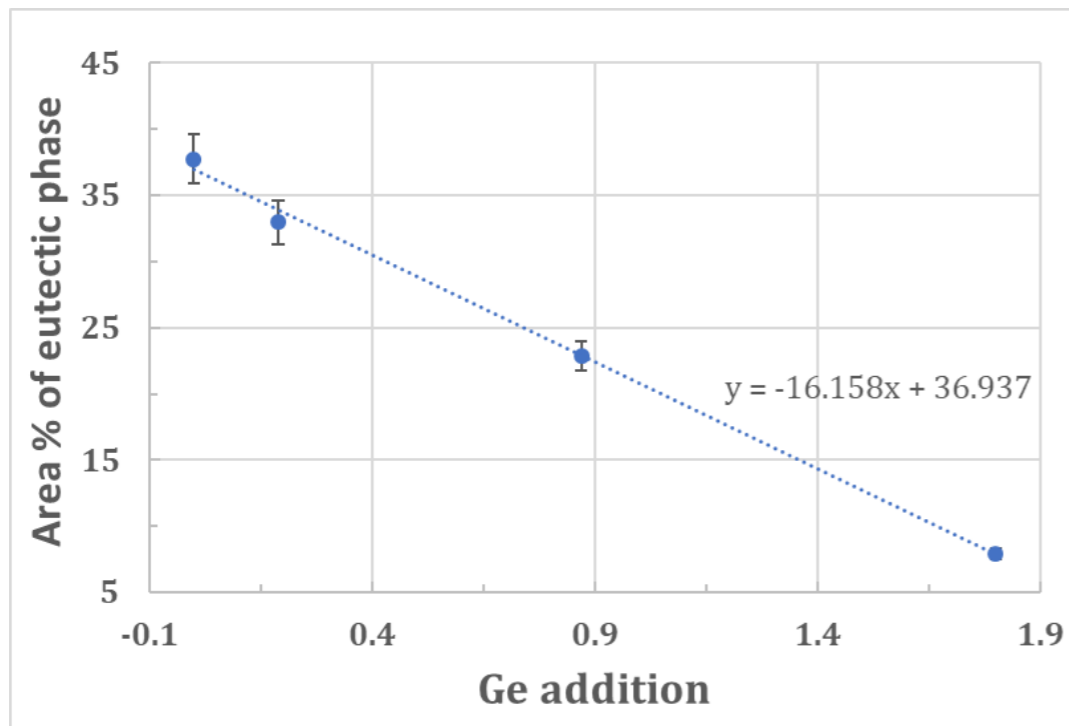
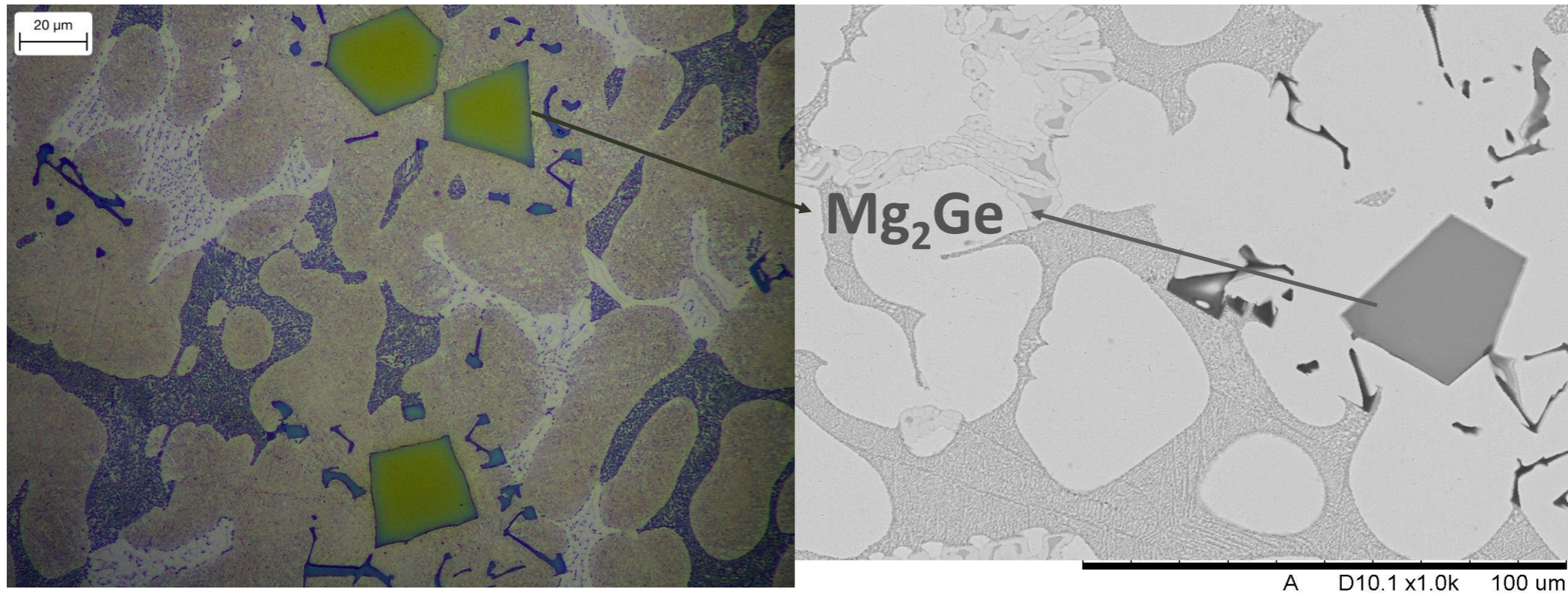
0.87 Ge



2018/08/09 12:35 A D10.1 x500 200 um

1.8 Ge

Volume fraction analysis:



SVET : Results

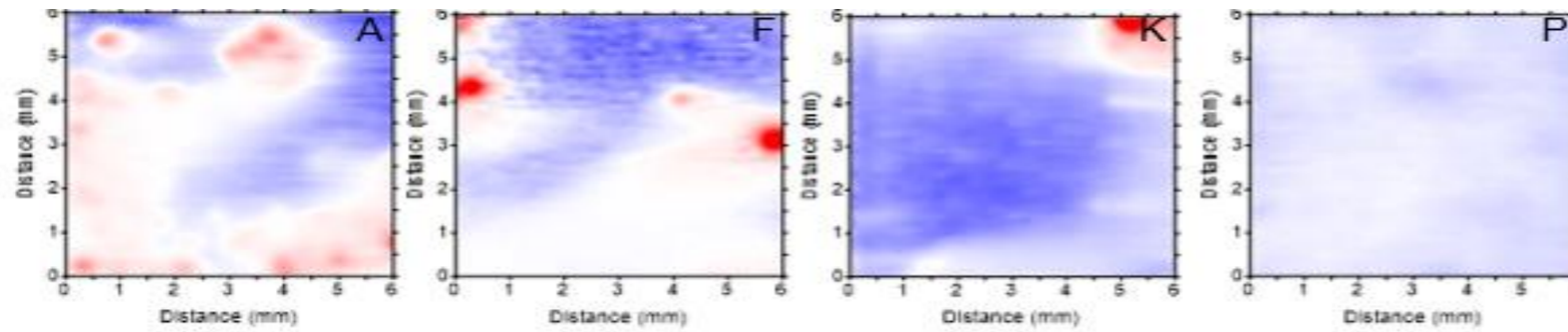
0 Ge

0.19 Ge

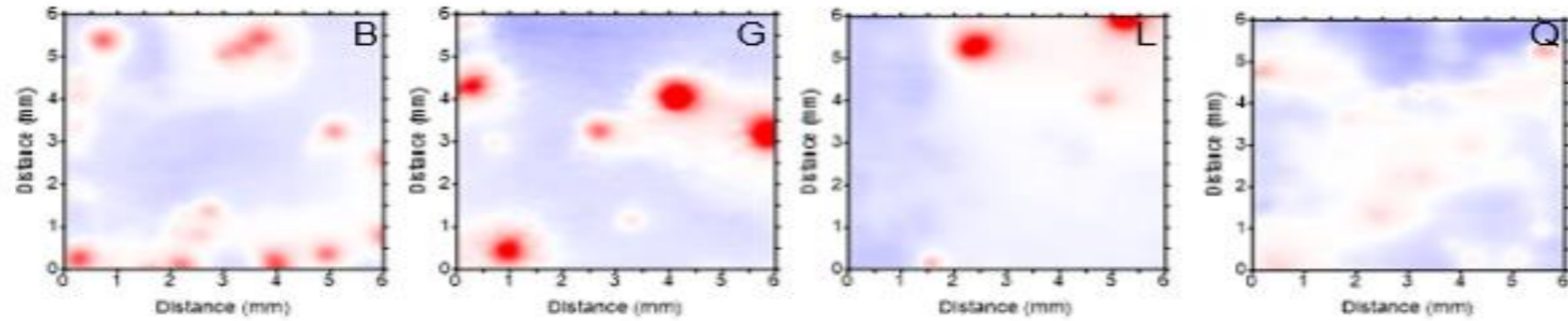
0.87 Ge

1.8 Ge

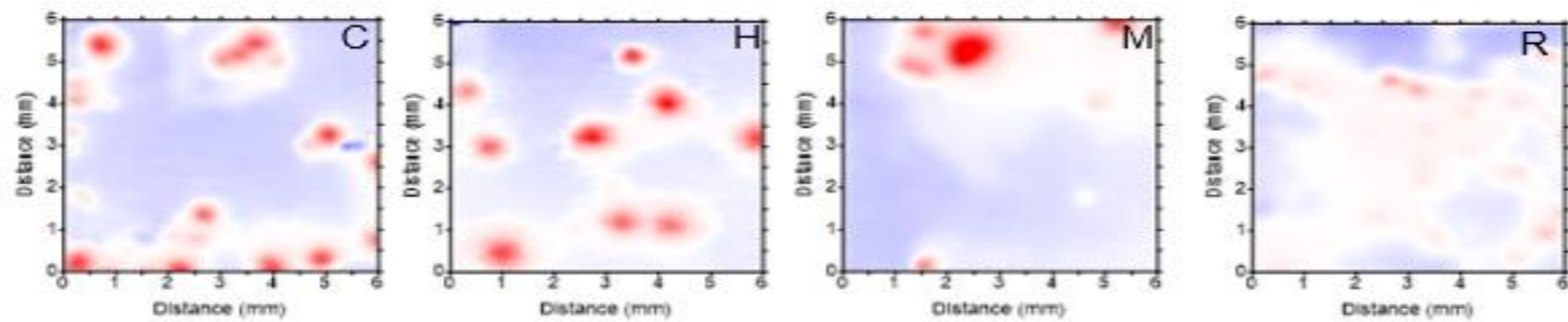
1 hour



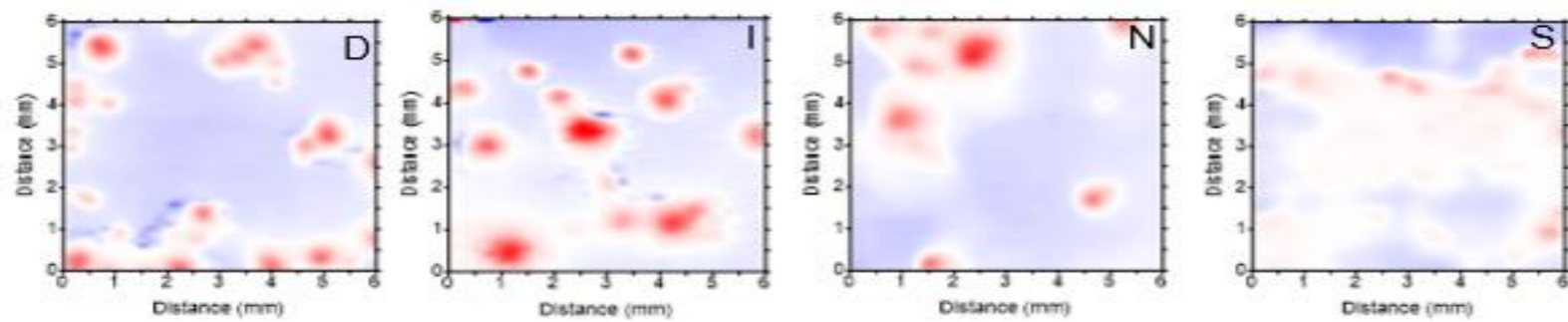
6 hours



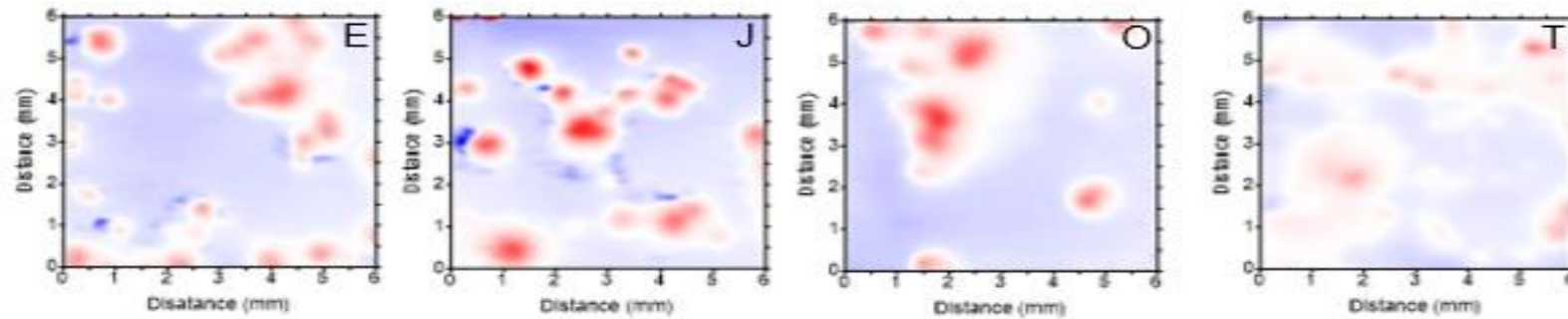
12 hours



18 hours



24 hours



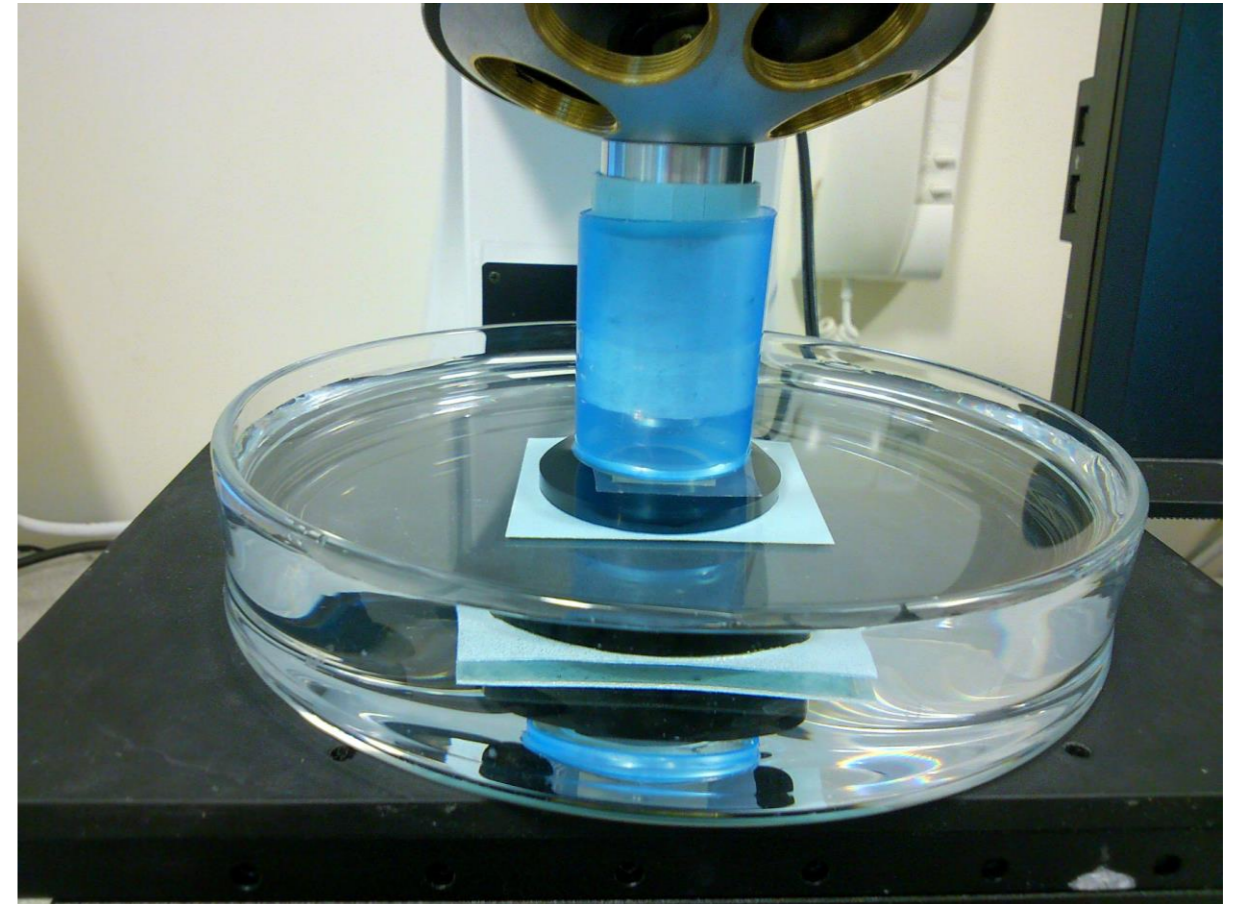
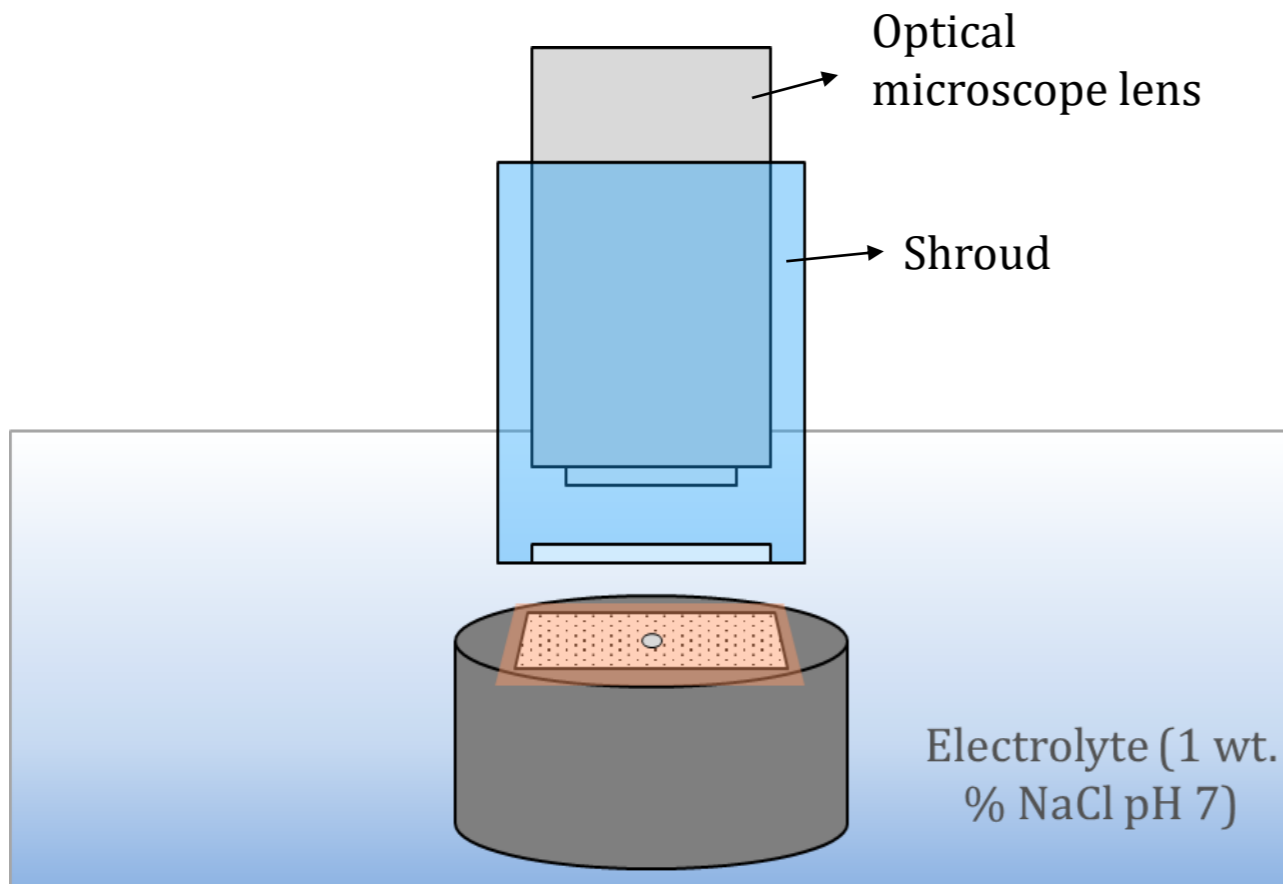
Sample	Mass Loss (g/m ²)
0 Ge	5.01 ± 0.49
0.19 Ge	8.53 ± 2.16
0.87 Ge	4.41 ± 0.51
1.80 Ge	2.11 ± 0.84



Electrolyte: 1 wt.% NaCl
pH 7

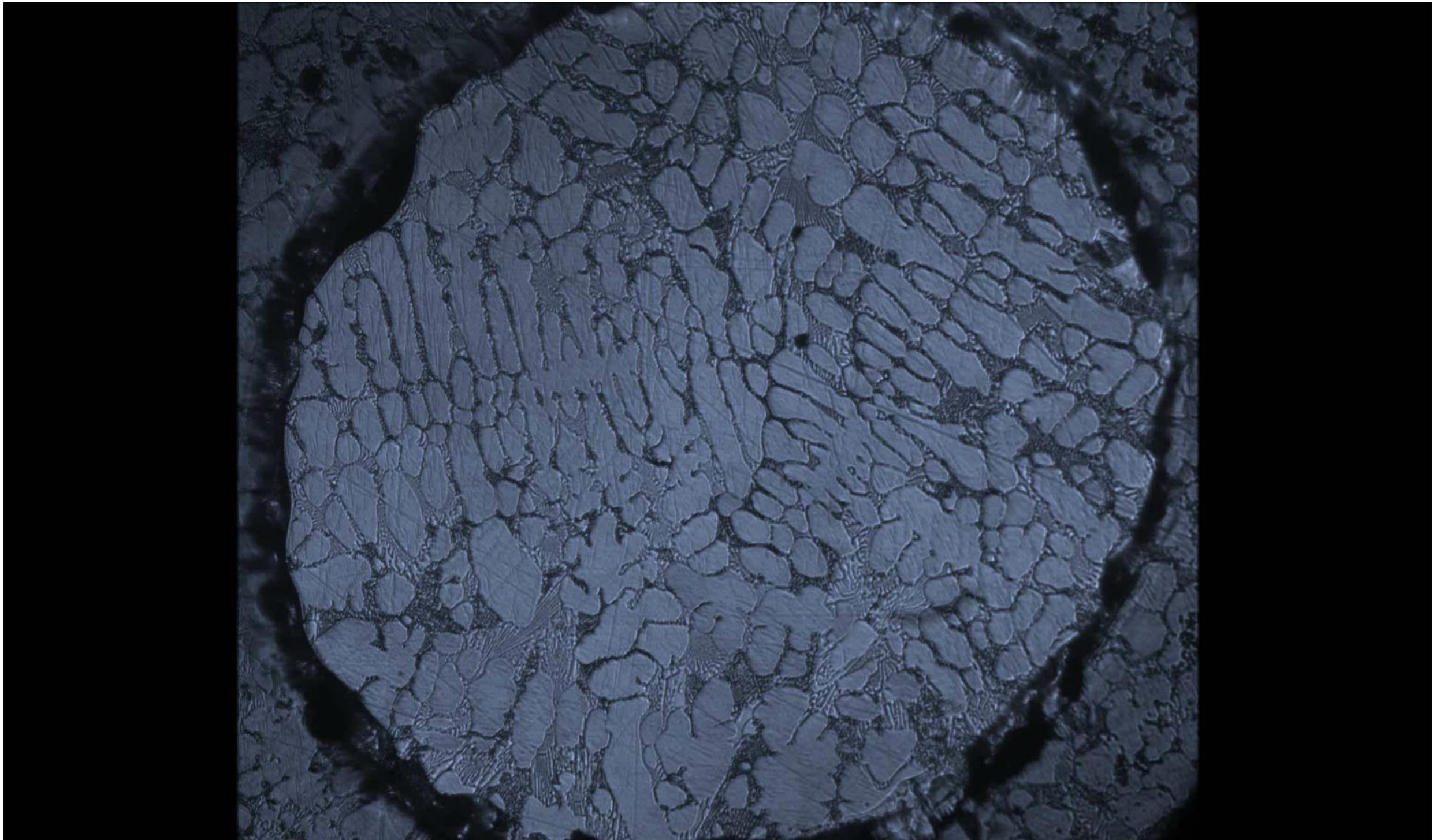
Time: 24 hours

Time-lapse Microscopy (TLM):



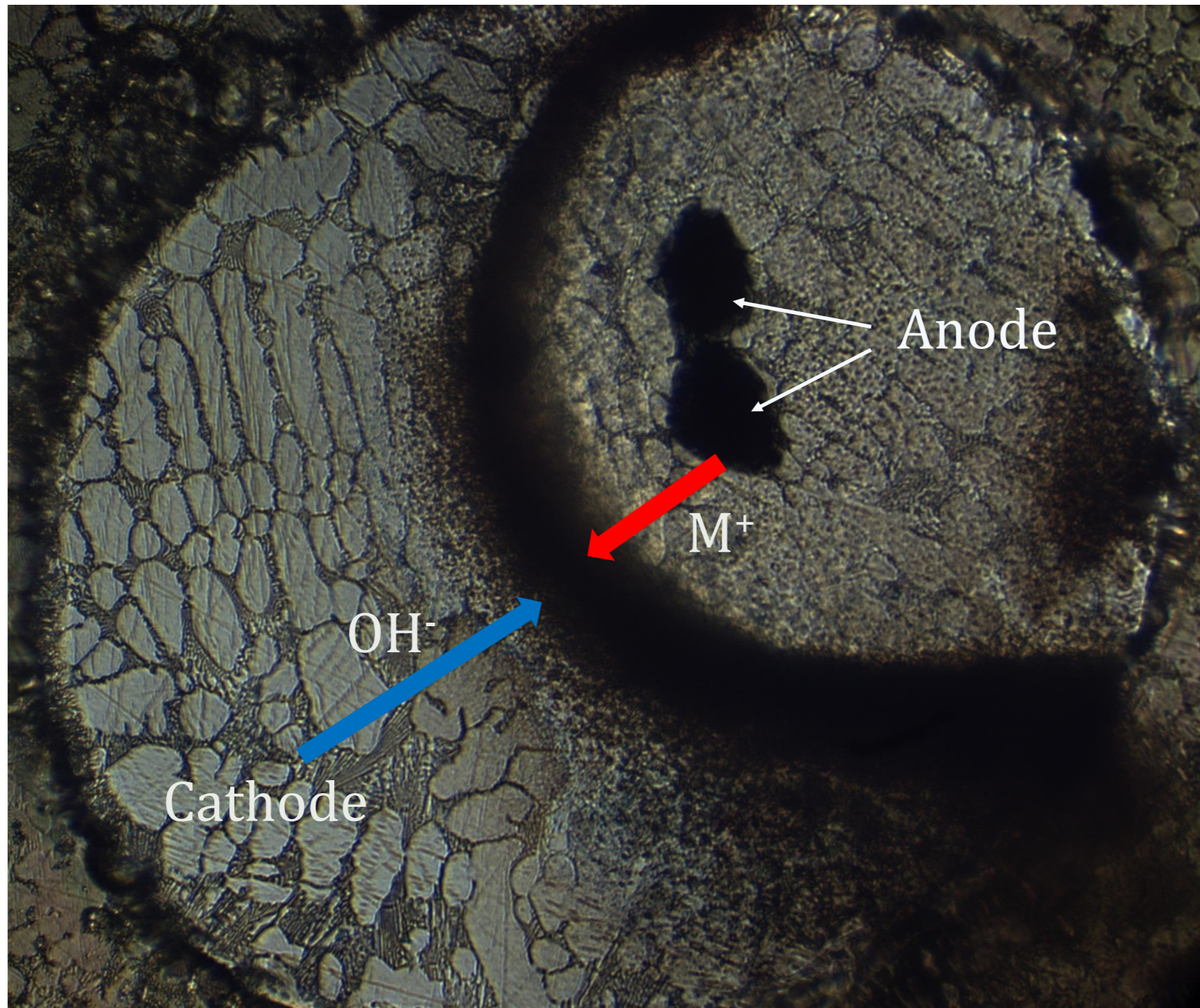
TLM: 0 Ge in 1 wt.% NaCl pH 7 for 24 hours

250 μm



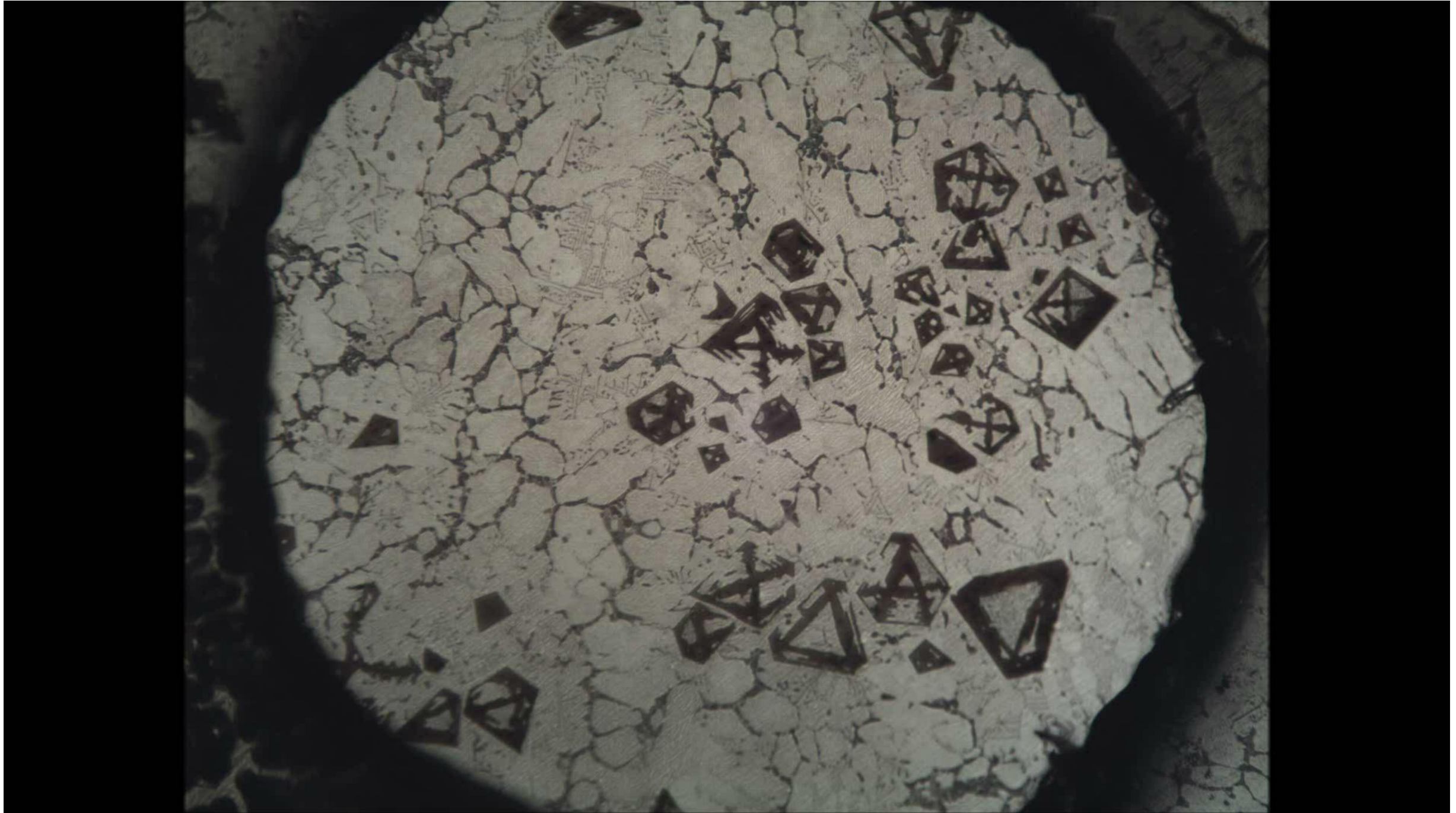
TLM: ZMA- 0 Ge

250 μm

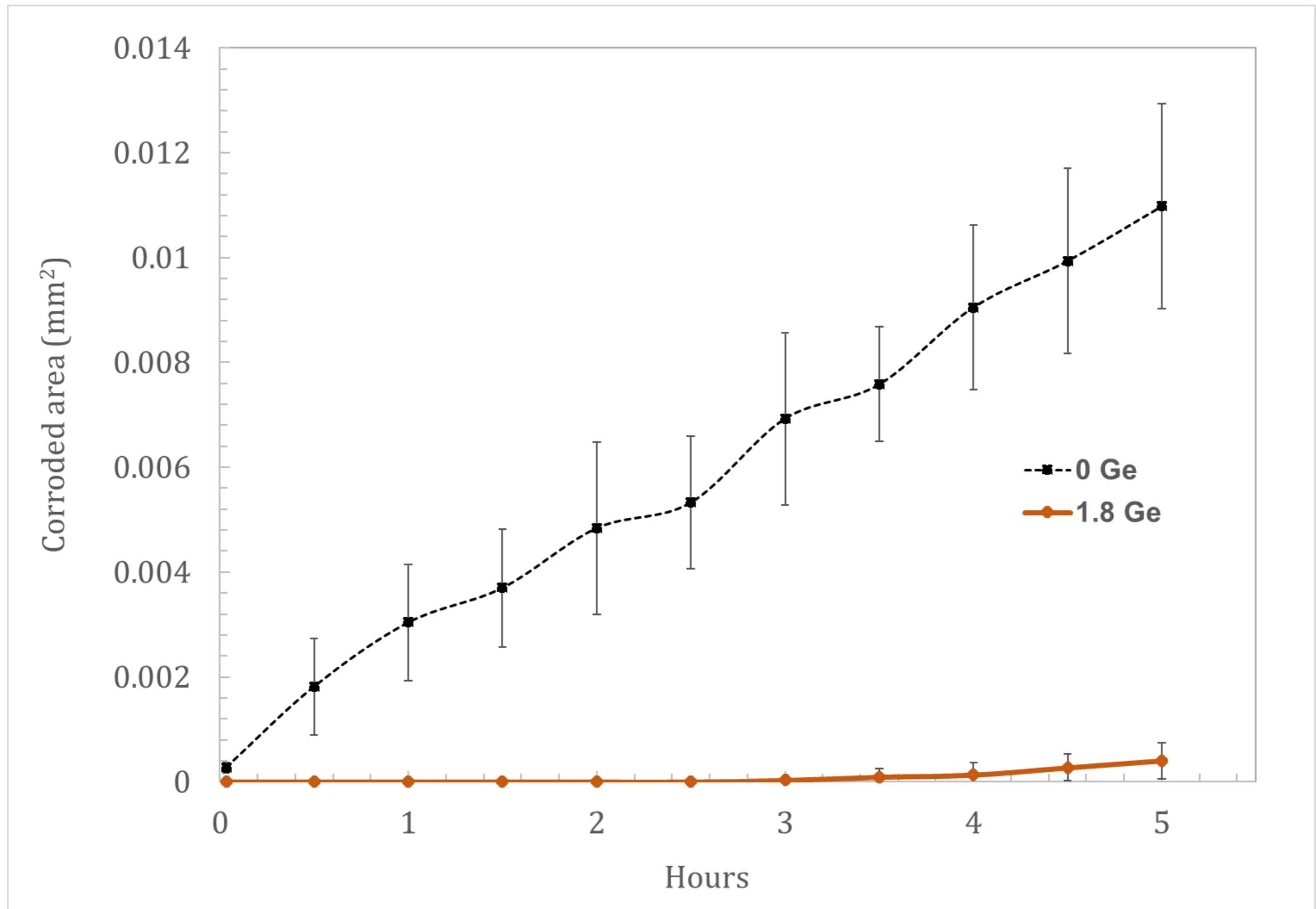


TLM: ZMA – 1.8 Ge in 1 Wt. % NaCl pH 7 for 24 hours

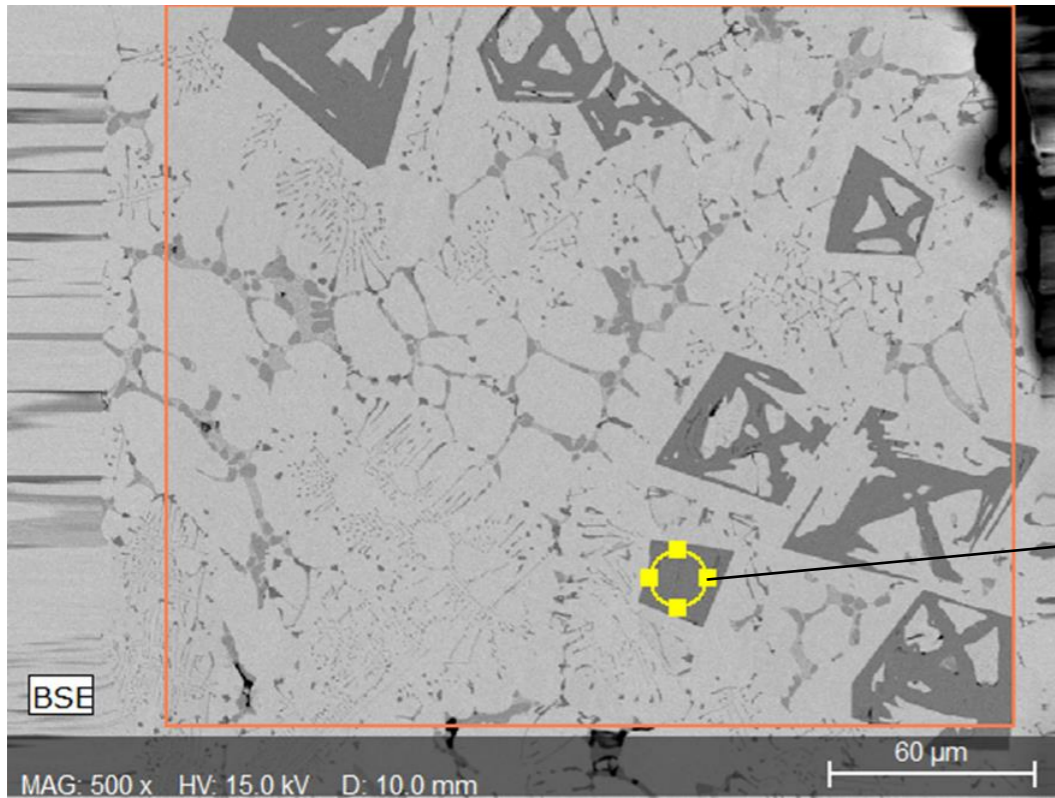
250 μm



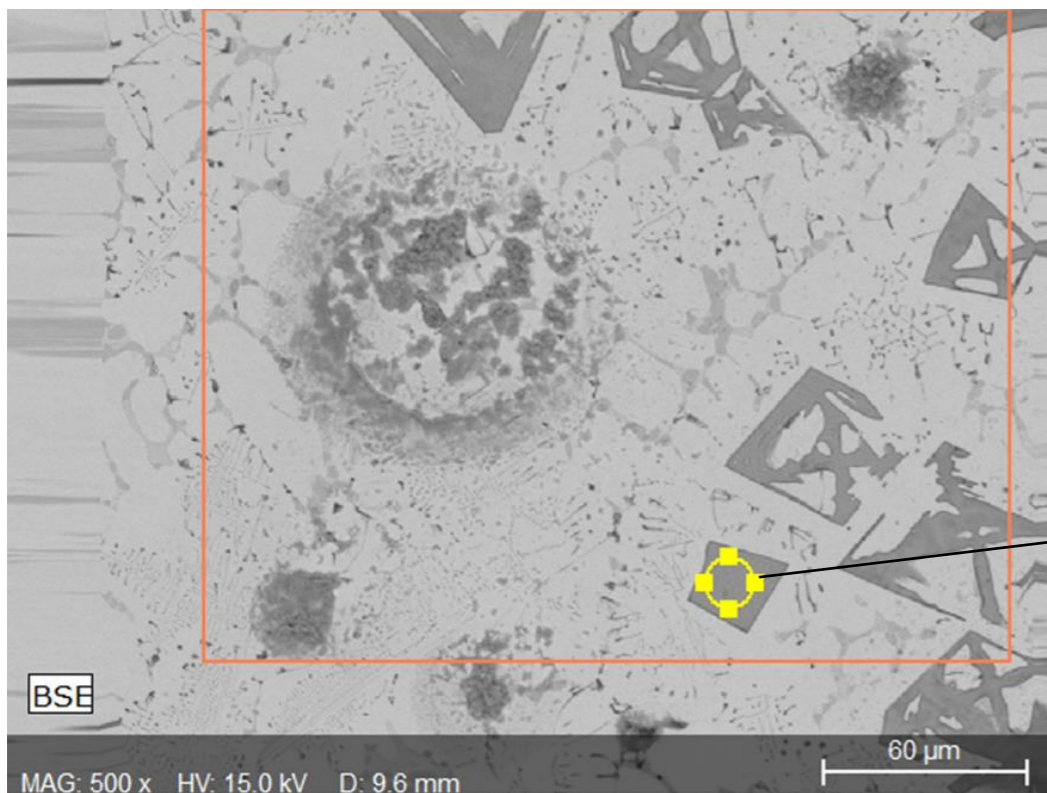
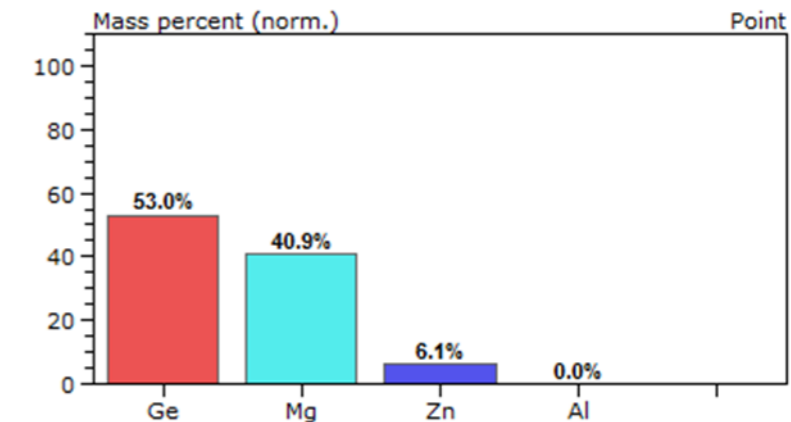
Corroded Area:



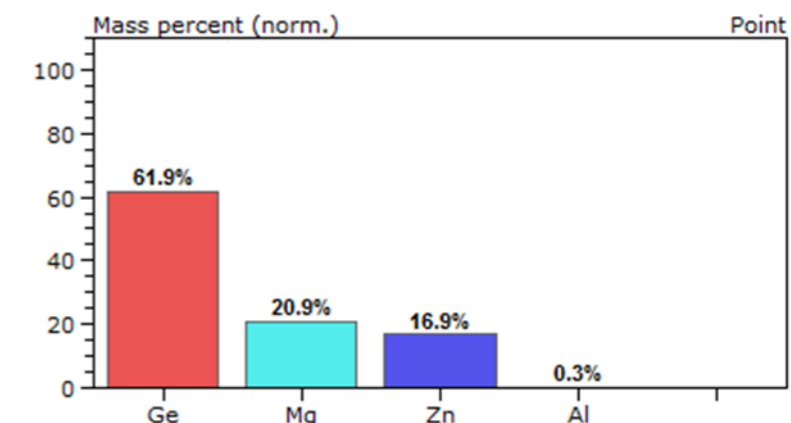
Scanning Electron Microscope (SEM): EDS analysis of Mg₂Ge crystal before & after 2 hours of immersion



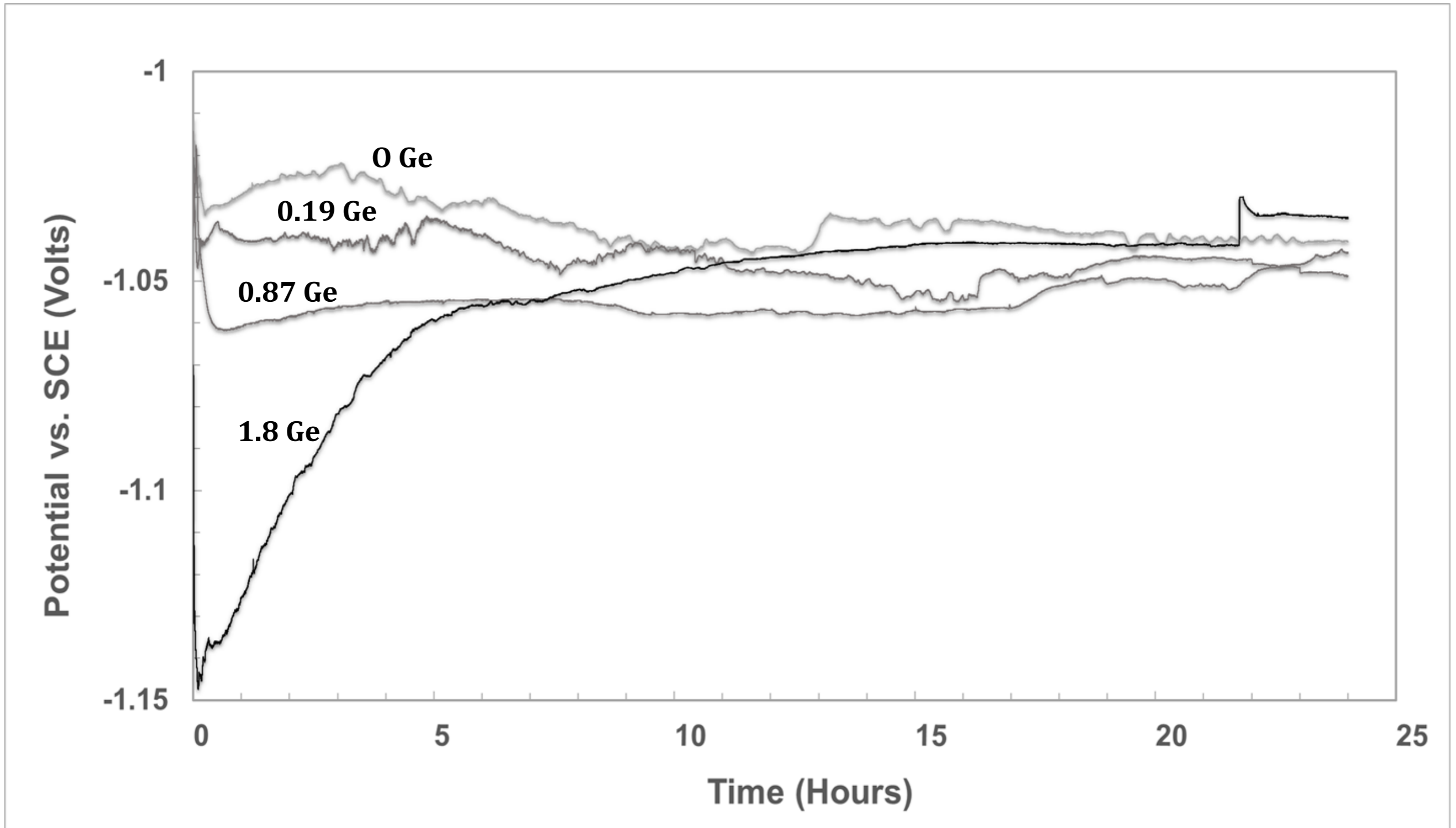
Element	Norm. C (wt. %)
Germanium	52.98
Magnesium	40.92
Zinc	6.09
Aluminium	0.00
Total	100.00



Element	Norm. C (wt. %)
Germanium	61.90
Magnesium	20.91
Zinc	16.88
Aluminium	0.31
Total	100.00

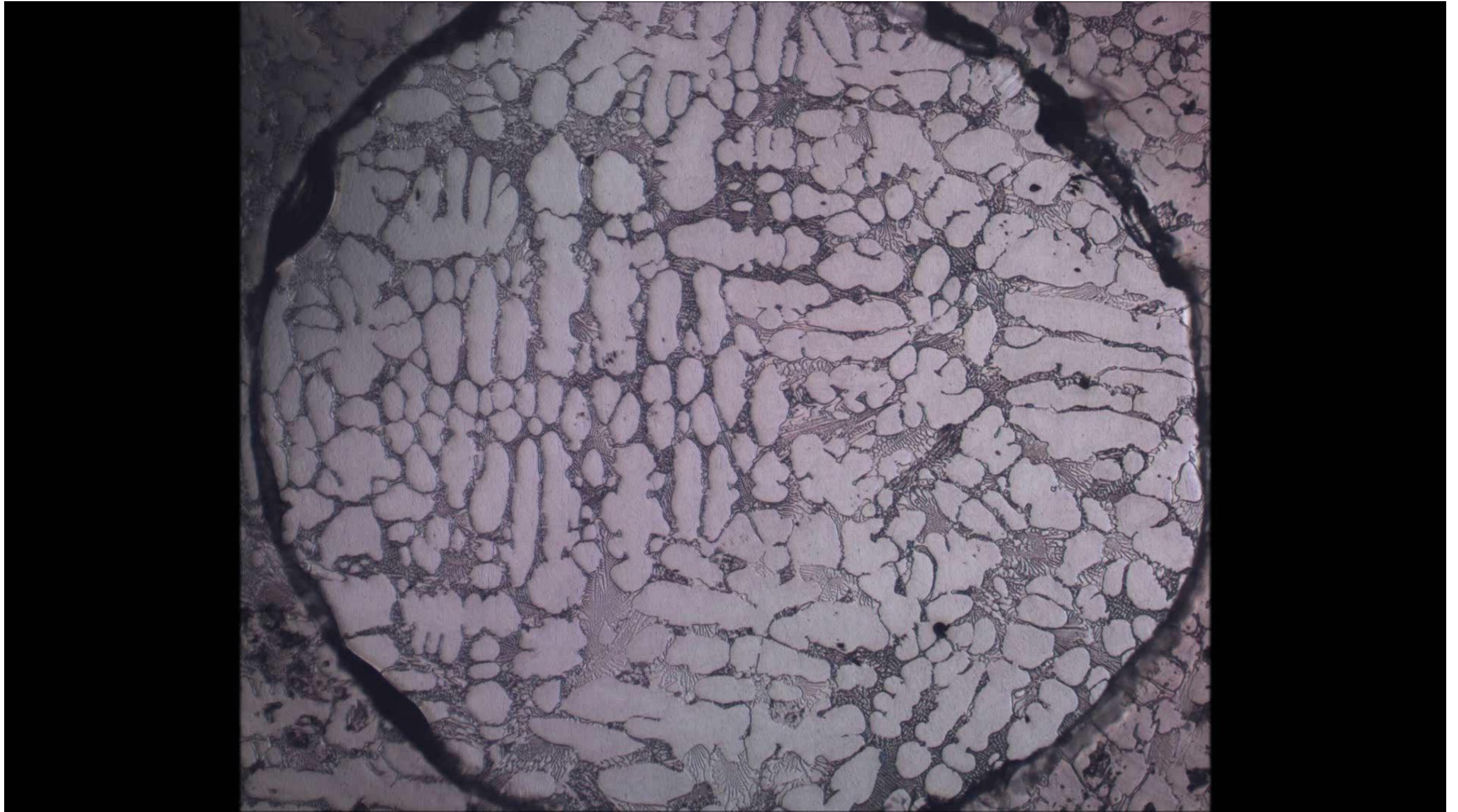


Electrochemical Measurements: OCP



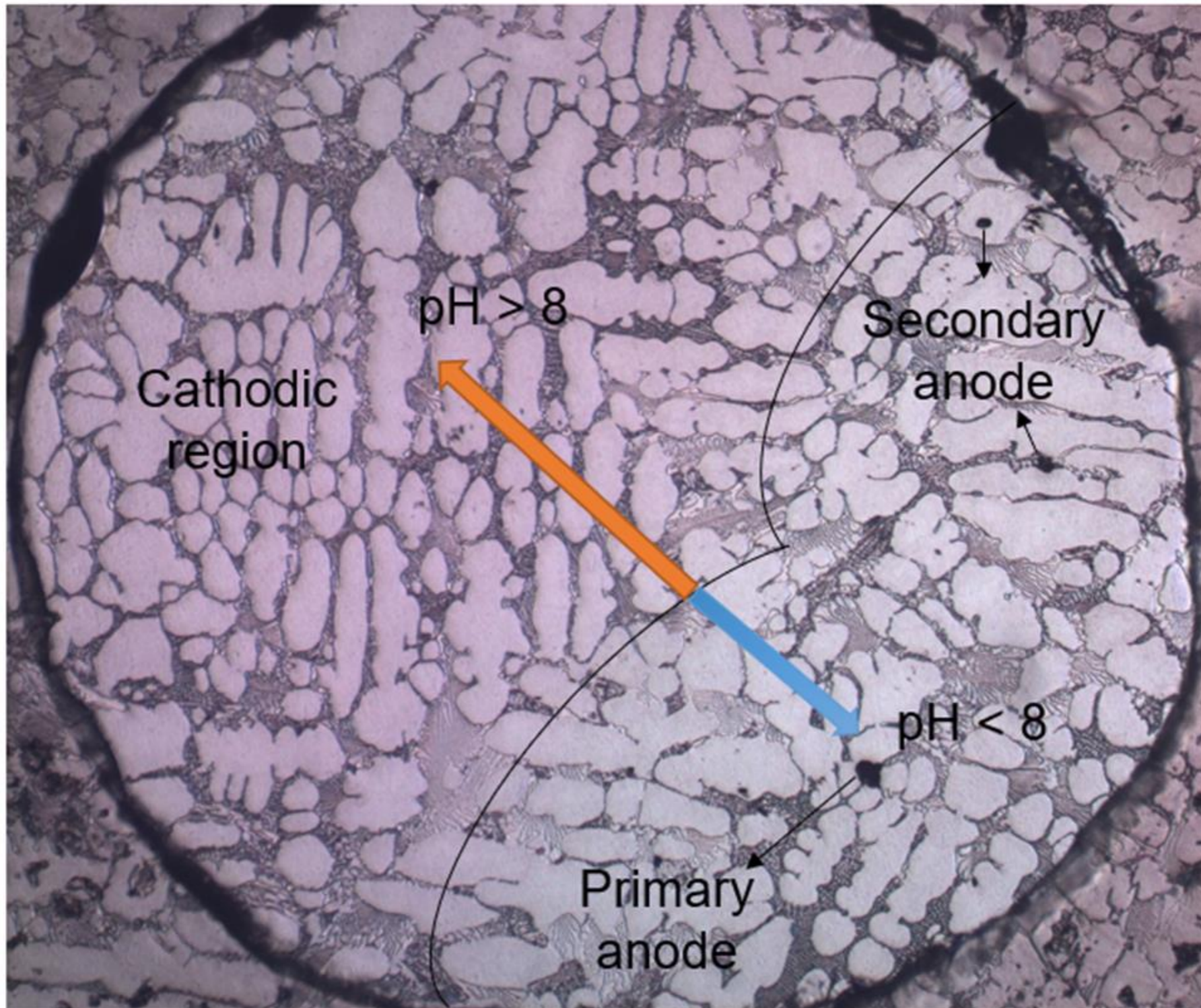
TLM: 0 Ge Indicator in 1 wt. % NaCl pH 7 with indicator for 24 hours

250 μm



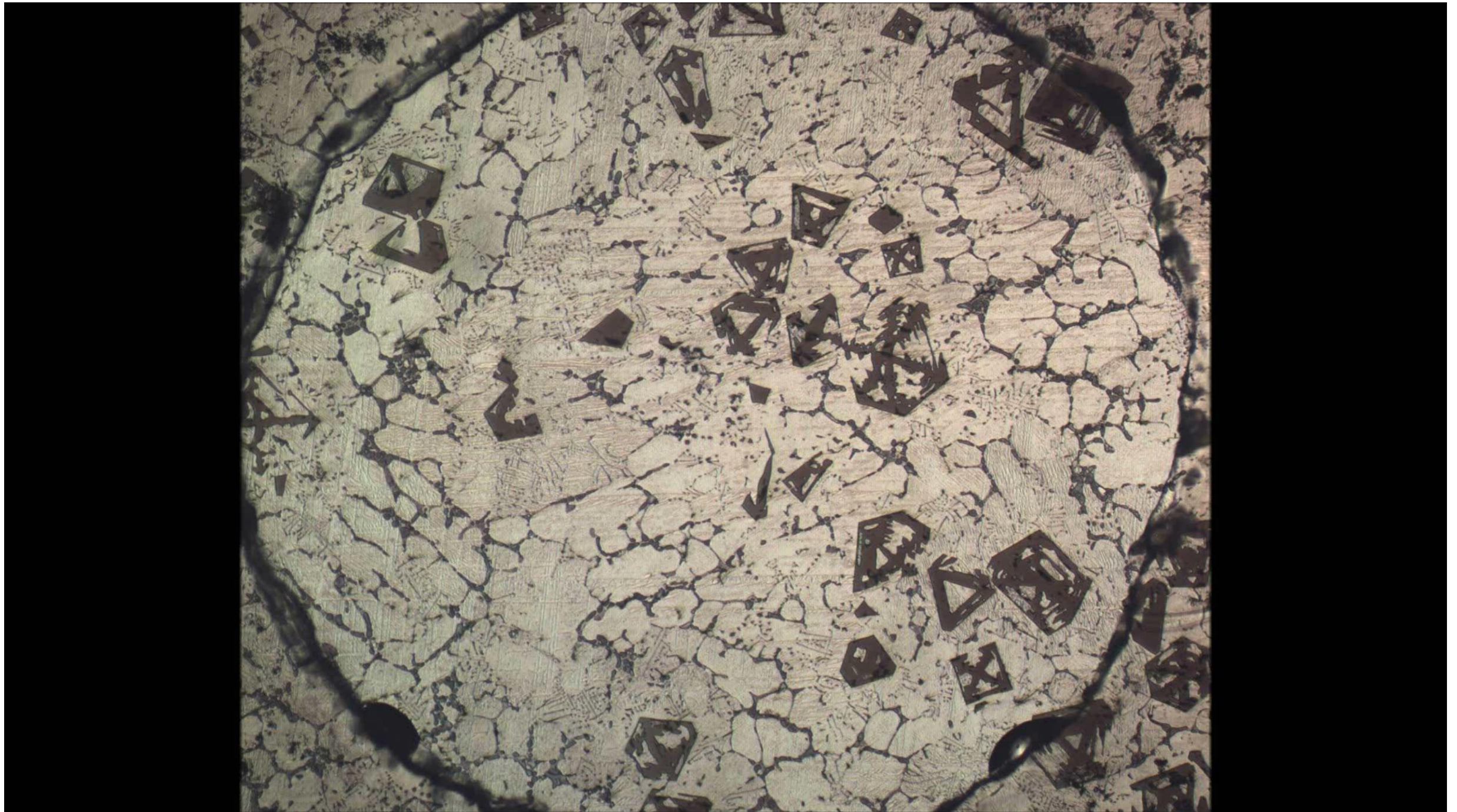
TLM: 0 Ge (Indicator) after 4 minutes

250 μm

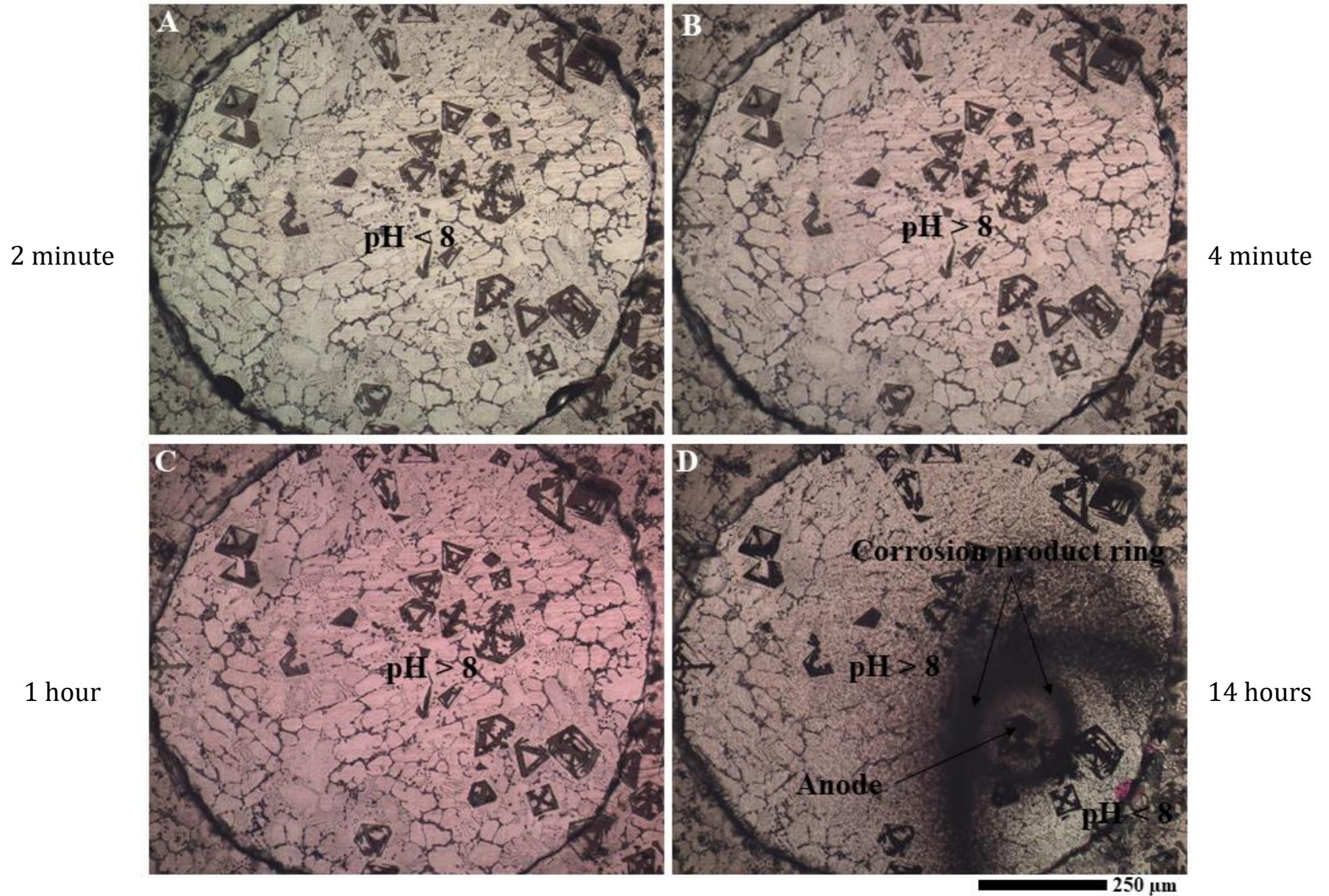


TLM: 1.8 Ge Indicator in 1 wt. % NaCl pH 7 with indicator for 24 hours

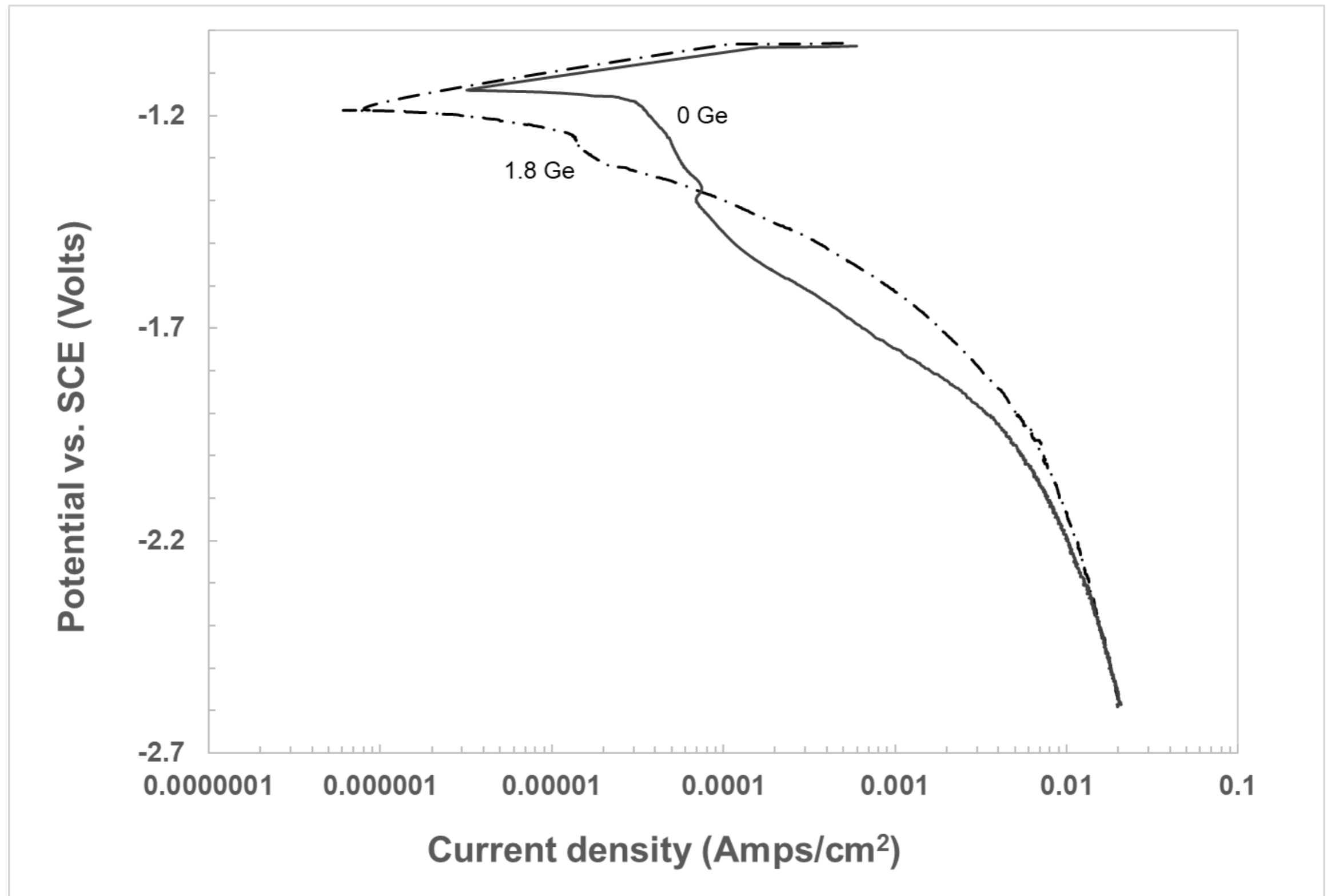
250 μm



TML: 1.8 Ge (Indicator)



Electrochemical Measurements: Cathodic polarisation



Conclusion:

- Leads to formation of Mg_2Ge crystals
- Mg_2Ge act as Mg^+ reservoir
- Mg^+ ions leads to increase in pH, consequently improving corrosion resistance.

Thank You

Prof. James Sullivan

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Prof. G. Williams

Tim Savill

